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The volume occupied by one mole of an ideal gas at 273 K and 1.01×10^5 Pa is 22.4 dm³.

What volume, in dm³, is occupied by 3.20 g O₂ (g) at 273 K and 1.01×10^5 Pa? A. 2.24 B. 4.48 C. 22.4 D. 71.7

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paper was okay though. Past papers really helped. P.S. Topic 2 question on the number of electrons in NUCLEUS was LIT.

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